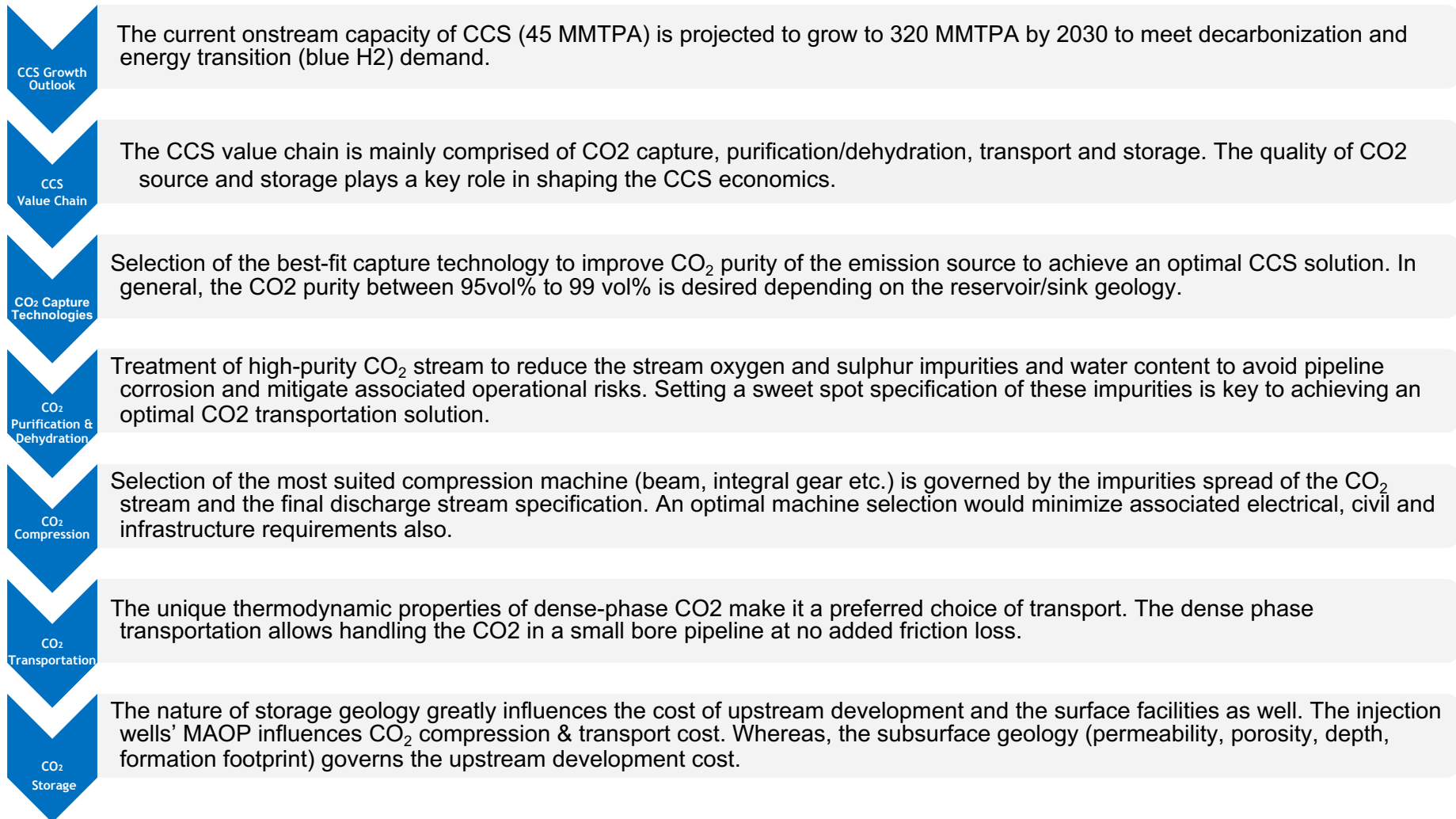
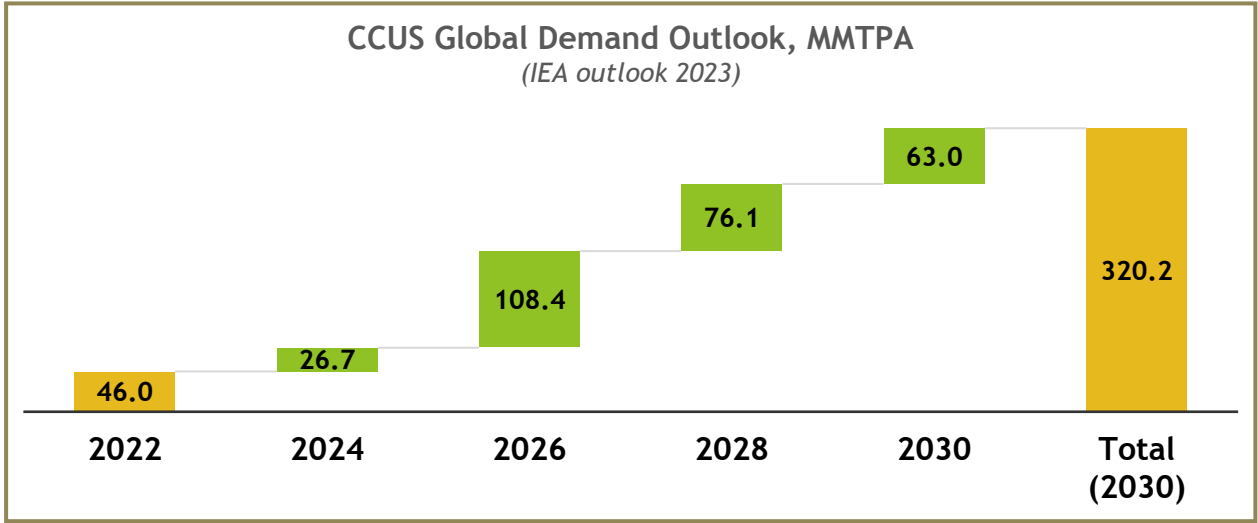
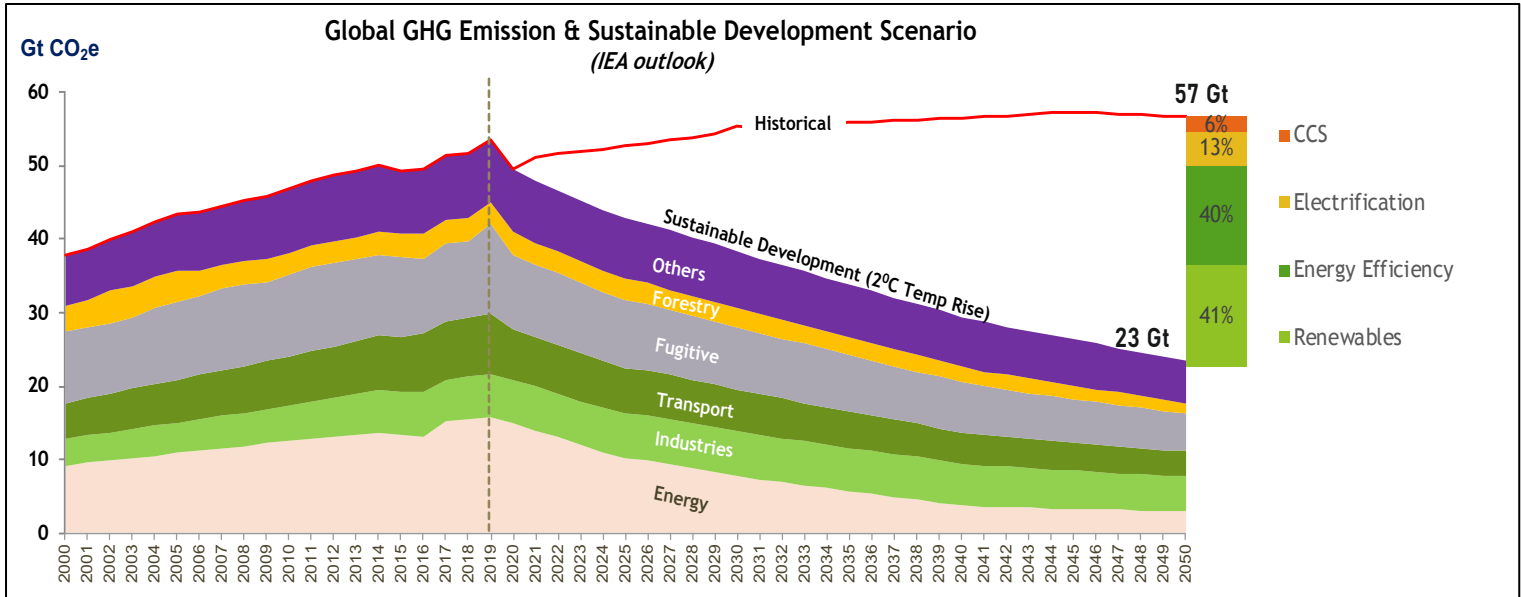




## **Introduction of Carbon Capture and Storage (CCS)**





- 6% share of low carbon energy (blue H<sub>2</sub>) in the global abatement portfolio to meet sustainable development scenarios.
- This reflects a dominant global growth outlook of CCS to facilitate energy transition.
- The second sketch projects an accelerated growth outlook for CCS

CCS Outlook

CCS Value Chain

CO<sub>2</sub> Capture

CO<sub>2</sub> Treatment

Comp & Dehy

Transportation

Storage



CO<sub>2</sub> Emission  
Source

Enhance CO<sub>2</sub>  
source purity  
(≥95 vol%)

Removal of  
oxygen, sulfur  
and water  
impurities

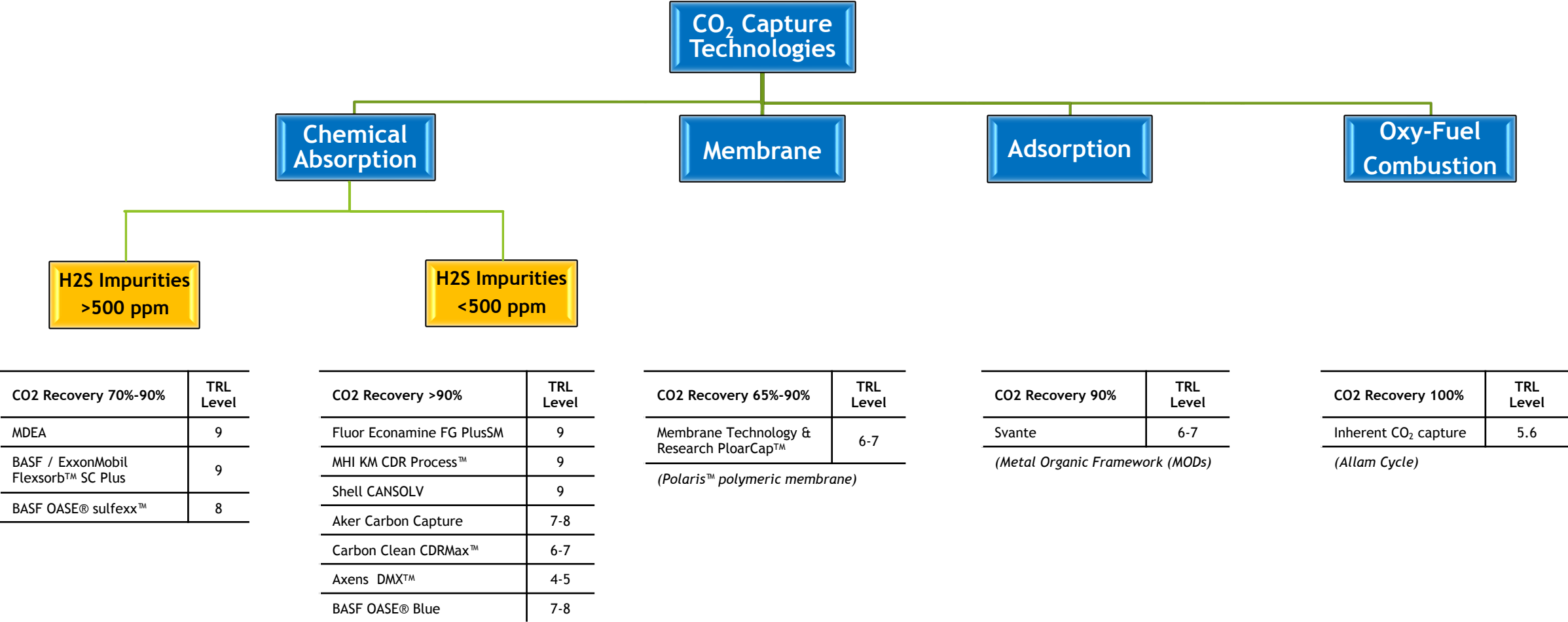
Conduit to  
deliver CO<sub>2</sub>  
molecules from  
source to sink

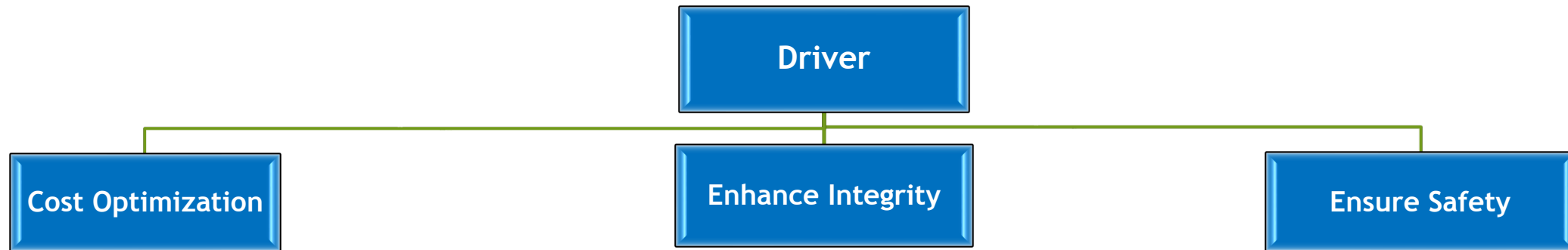
Store CO<sub>2</sub>  
molecules safely  
and permanently

The value chain definition assists in defining the following:

- CCS development's cost benchmarking with global developments
- Develop an effective proposal to obtain government incentive
- Help to form a viable JV having clarity on scope boundaries and responsibilities

- A wide range of emitters spread on CO<sub>2</sub> purity.
- The cost of capturing CO<sub>2</sub> decreases as the emission source's CO<sub>2</sub> purity increases.
- Emission sources with <4 vol% CO<sub>2</sub> purity & <0.3 MMPTA capacity incur very high capture costs and shall be avoided in the development.
- High cost of capture for emitters having H<sub>2</sub>S impurities > 500 ppm due to availability of limited capture technologies and poor capture performance.
- Efforts are underway to reduce capture costs <\$40/ton





Treating oxygen, sulfur and water impurities of the CO<sub>2</sub> stream to a sweet spot lowers the development cost by:

- Requiring less stringent pipeline material specification
- Improving plant availability and reducing the risk of unplanned shutdowns.
- Simplifying CCS configuration
- Allowing for dense phase CO<sub>2</sub> transport

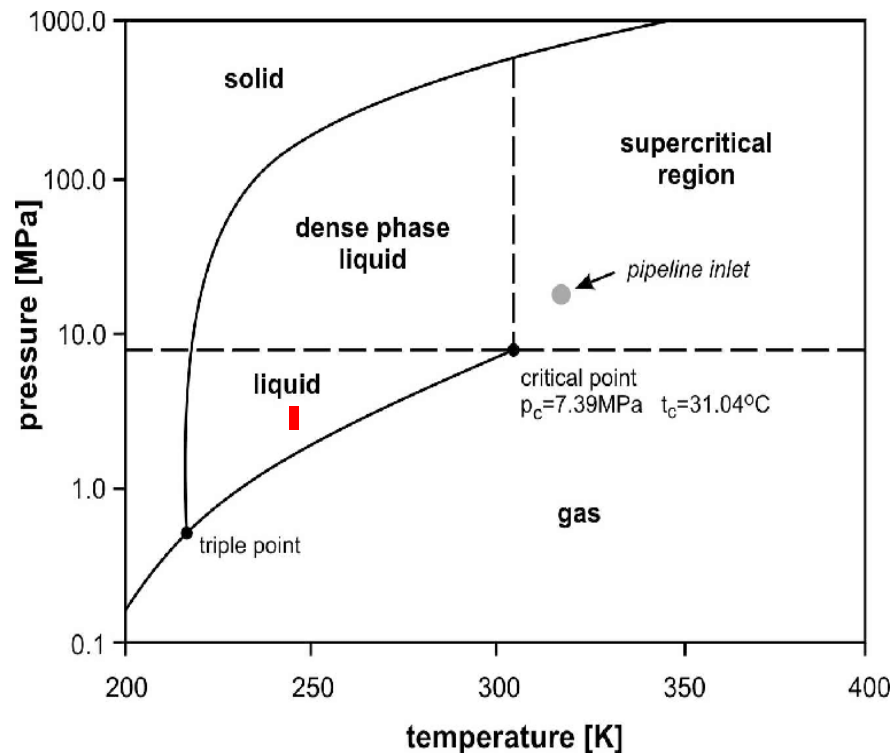
Limiting the presence of oxygen, SO<sub>x</sub>, NO<sub>x</sub>, H/C impurities to a safe level in a CO<sub>2</sub>-rich environment improves the integrity of the system by:

- Reducing the risk of water and hydrocarbon dropouts in CO<sub>2</sub>-rich gas phase environment to prevent corrosion
- Eliminate the risk of strong acid formation in a CO<sub>2</sub> rich environment due to the presence of SO<sub>x</sub> and NO<sub>x</sub> in the presence of O<sub>2</sub>.

Treated CO<sub>2</sub> stream eliminates the risk of corrosion in gas phase transport, allowing the routing of a gas phase pipeline through populated areas to reduce the risk of fatalities in the event of an uncontrolled bulk release of CO<sub>2</sub>.

- CO<sub>2</sub> is 1.5 times heavier than air
- Exposure to a 3vol% concentration of CO<sub>2</sub> for one hour can cause headaches, increased respiratory and heart rate, dizziness, muscle twitching and confusion.
- Just one minute of exposure to a 15vol% concentration of CO<sub>2</sub> can pose an immediate threat to life.

Chart-1: CO<sub>2</sub> Phase Envelop



- An ideal operating range for the CO<sub>2</sub> cryogenic process is shown in red on chart-1 (Pressure: 3.8 -5 Mpa, and temp 223°K/-50°C)
- It's important to note that the quality specifications for liquid CO<sub>2</sub> can increase costs..
- The cryogenic option offers lower operating expenses (OPEX) compared to the compression option because of its lower power consumption. However, the cryogenic option requires higher initial capital expenditure (CAPEX) and involves more operational complexities, making it less attractive than compression.
  - The cryogenic option necessitates upfront removal of heavy hydrocarbons and water to prevent fouling in the cold box at -5°C.
  - Additionally, there's a risk of potential loss of CO<sub>2</sub> due to the venting of non-condensable gases in the cryogenic process.





- 580-650 psi pressure range offers minimum saturation water in the CO<sub>2</sub> stream, i.e. a sweet operating spot for the dehydration unit delivering an optimal compression and dehydration solution.
- TSA (temperature swing adsorber) is preferred over the legacy TEG option due to its following technical superiority.
  - Achieves a stringent water specification required to avoid corrosion caused by acid impurities in the CO<sub>2</sub> stream.
  - Avoid the presence of TEG carryover in the treated stream to CO<sub>2</sub> pipeline from to severe corrosion
- A corrosion study conducted using premier software such as OLI suggests that even a small presence of TEG in a dry CO<sub>2</sub>-rich environment could cause severe corrosion in the presence of parts per million levels of acid impurities.

Parameters	TEG	TSA
Sweet Operating Point	Approx. 580 psi	Approx. 580 psi
Operating Temp Limit	≥ 100F (foaming below this temp)	≥ 75F (Hydrate formation issue)
Achievable Water Spec	≥ 150 ppm	≥ 40 ppm
Water Spec to avoid hydrate formation	50	50
TEG carryover to avoid corrosion	<10 ppb	NA
CAPEX/OPEX	Dense phase	All
Preferred Service	Dense phase	All

# Best Fit Compressor Type for CO<sub>2</sub> Service

CCS Outlook

CCS Value Chain

CO<sub>2</sub> Capture

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Comp & Dehy

Transportation

Storage

Parameters	Beam Type	Integral Gear (IG)	Notes
CAPEX	-	Lower	IG offers a compact footprint by capitalizing on a water-cooled interstage option.
Acceptable H <sub>2</sub> S impurity in CO <sub>2</sub> stream	No limit	16 ppm (max)	IG requires expensive tandem/double dry gas seals for CO <sub>2</sub> service with more than 16 ppm H <sub>2</sub> impurities, posing added risk of frequent seal failure.
Interstage Cooling	Air-cooled or water cooled	Water cooled recommended	IG offers reduced footprint and lower power consumption with water-cooled option
Driver Type	No reservation	No reservation	Driver selection (VFD/VSG/Fixed speed) is a function of motor size (MW) and service type (turndown ratio and frequency)

- In general, IG compressors are more economical than beam compressors
- IG is not recommended on the technical ground for CO<sub>2</sub> streams having H<sub>2</sub>S impurity >16 ppm

- The unique thermodynamic properties of dense phase (supercritical) CO<sub>2</sub> make it a preferred choice of transport.
  - The dense phase demonstrates a viscosity similar to that of a gas, but with a density closer to that of a liquid, allowing for the use of a small bore pipeline.
  - The dense phase has excellent solvent properties, significantly reducing the risk of corrosion in the pipeline.
- Dense phase operation over 1,400 psig is desired to avoid inflection of the density curve (irregular pressure drop profile).
- The presence of non-condensable gases such as hydrogen, methane, and nitrogen in CO<sub>2</sub> reduces the density of the dense phase, leading to an increase in frictional pressure drop.

- The geology of sinks greatly influences the development cost. The following supports a low-cost CCS development.
  - Lower WHIP offers overall lower surface development cost.
  - High permeability/porosity offers less number of injection wells
  - Shallow sinks offer low upstream development costs.
  - Circular or square formation of reservoirs reduces infrastructure costs associated with wellhead (power, flowlines etc.)
- Sinks close to emitters reduce CO<sub>2</sub> compression and transport costs.